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1 **Deep Atlantic Ocean carbon storage and the rise of 100,000-year glacial cycles**

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Over the past three million years, Earth's climate oscillated between warmer interglacials with reduced terrestrial ice volume and cooler glacials with expanded polar ice sheets. These climate cycles, as reflected in benthic foraminiferal oxygen isotopes, transitioned from dominantly 41-kyr to 100-kyr periodicities during the mid-Pleistocene (1,250 to 700 ka). Because orbital forcing did not shift at this time, the ultimate cause of this mid-Pleistocene transition (MPT) remains enigmatic. Here we present foraminiferal trace element (B/Ca, Cd/Ca) and Nd isotope data that demonstrate a tight linkage between Atlantic Ocean meridional overturning circulation and deep-ocean carbon storage across the MPT. Specifically, between 950 and 900 ka, carbonate ion saturation decreased by 30 $\mu\text{mol/kg}$ and phosphate concentration increased by 0.5 $\mu\text{mol/kg}$ coincident with a 20% reduction of North Atlantic Deep Water contribution to the abyssal South Atlantic. These results demonstrate that the glacial deep Atlantic carbon inventory increased by approximately 50 gigatons during the transition to 100-kyr glacial cycles. We suggest that the coincidence of our observations with evidence for increased terrestrial ice volume reflects how weaker overturning circulation and Southern Ocean biogeochemical feedbacks facilitated deep ocean carbon storage, which lowered atmospheric pCO_2 and thereby enabled expanded terrestrial ice volume at the MPT.

Cyclic glaciations are the primary feature of Earth's climate since the late Pliocene and occur at periodicities linked to variations in solar insolation¹. However, the dominant periodicity of glaciations transitioned from 41-kyr to 100-kyr during the mid-Pleistocene without concomitant changes in external insolation forcing²⁻⁵. It has been

suggested that the emergence of high-amplitude 100-kyr cycles in surface and deep ocean climate records during the MPT^{4,5,6} relates to internal climate amplifiers (“feedbacks”)³, including changes in ice sheet dynamics^{3,5,6-11}, ocean circulation^{12,13}, and interactions of these processes with Earth’s carbon cycle¹⁰⁻¹⁴. Although temporal resolution is low, atmospheric pCO₂ reconstructions across the MPT show that glacial pCO₂ decreased by ~30 µatm sometime between 1,000 and 800 ka^{11,15,16}, consistent with more voluminous glacial ice sheets^{10,17}. The missing glacial pCO₂ was most likely sequestered in the deep ocean¹⁸ as suggested by benthic foraminiferal carbon isotope records ($\delta^{13}\text{C}_b$)^{5,14,19}. However, evidence from $\delta^{13}\text{C}_b$ reflects a combination of ocean circulation, air-sea gas exchange, and ocean carbon content, which complicates quantitative reconstructions of any one parameter^{13,14,19}. Because reliable quantitative records of ocean carbonate chemistry are sparse, it is debated how the ocean sequestered additional CO₂ at the MPT^{11,13}. Stronger quantitative constraints on deep ocean dissolved inorganic carbon (DIC) and its relation to meridional overturning circulation (MOC) are needed to evaluate whether, and by which mechanisms, the ocean stored additional carbon at the MPT.

Here we test the hypothesis that the onset of 100-kyr glacial cycles during the MPT was facilitated by enhanced deep ocean carbon storage accommodated by reduced Atlantic MOC^{12,13}. We present the first paired reconstructions of deep ocean DIC content and MOC using foraminiferal trace element (B/Ca, Cd/Ca) and Nd isotope proxies, recorded in sediments from Ocean Drilling Program Site 1267 from the Southeast Atlantic Ocean (Fig. 1). In combination with previously published MOC and trace element reconstructions, our results indicate coeval MOC reduction and increased deep Atlantic DIC after 950 ka. We evaluate the contribution of our DIC estimates to

atmospheric pCO₂ based on ocean CO₂ sequestration mechanisms identified by modeling studies of the Last Glacial Maximum (LGM, ~20 ka), and hypothesise a sequence of climatic events that link changes in ocean circulation, carbon cycle and Earth's cryosphere with the emergence of 100-kyr glacial cycles during the MPT.

Carbonate chemistry, nutrients, and MOC across the MPT

Site 1267 is located at 4.4 km water depth in the Angola Basin north of Walvis Ridge²⁰, within dense Atlantic Ocean bottom waters (Fig. 1b). These waters comprise two general endmembers with characteristic chemical properties: North Atlantic Deep Water (NADW) feeds the basin from the north, and Antarctic Bottom Water (AABW) from the south. High oxygen, low-CO₂ and nutrient-poor NADW carries a low Nd isotope ratio (expressed as more negative ϵ_{Nd} , Methods), while low oxygen, high-CO₂ and nutrient-replete AABW carries a more positive ϵ_{Nd} signature from admixed Pacific waters (Fig. 1). Because Walvis Ridge blocks AABW inflow from the south, AABW enters the Angola Basin via the Romanche Fracture zone (0°N), thereby limiting the contribution of the generally deeper AABW to abyssal Angola Basin waters today (Fig. 1b). However, Site 1267 exhibits higher nutrient conditions (indicated by low $\delta^{13}C_b$ values) during Pleistocene glaciations^{21,22}, suggesting that this location tracks the balance of NADW and AABW contributions to the deep Atlantic across the MPT.

The B/Ca and Cd/Ca ratios of the epibenthic foraminifer *Cibicidoides wuellerstorfi* reflect the deep-sea carbonate ion saturation state²³ ($\Delta[CO_3^{2-}]$) and phosphate concentration²⁴ ($[PO_4^{3-}]$), respectively (Methods). Changes in $\Delta[CO_3^{2-}]$ and $[PO_4^{3-}]$ are linked to changes in DIC through ocean biogeochemistry, and we apply these

proxies from Site 1267 alongside published B/Ca and Cd/Ca records from North Atlantic Site 607^{13,25} to quantitatively constrain deep Atlantic DIC across the MPT. Additionally, ϵ_{Nd} of planktic foraminiferal Fe-Mn oxide coatings at Site 1267 records the local expression of changes in NADW and AABW contributions to the deep Atlantic Ocean and complements a more southerly reconstruction of MOC strength across the MPT¹².

Site 1267 geochemical data show pronounced shifts in circulation and chemical properties between Marine Isotopic Stages (MIS) 26 (970 ka) and 22 (880 ka) (Fig. 2). ϵ_{Nd} suggests an ~20% reduction in NADW contribution to the South Atlantic between MIS 26/25 and MIS 23/22 (Methods), in agreement with records from Sites 1088 and 1090 located ~10° to the south¹² (Supplementary Fig. 1). These combined ϵ_{Nd} records indicate that the MOC perturbation between MIS 26 and 22 exceeds any previously studied glacial transition before 950 ka. This MOC reduction corresponds with abrupt increases in deep water corrosivity and nutrient content as indicated by an ~30 $\mu\text{mol/kg}$ $\Delta[\text{CO}_3^{2-}]$ decrease and an ~0.5 $\mu\text{mol/kg}$ $[\text{PO}_4^{3-}]$ increase at Site 1267 (Fig. 2c, d). This chemical transition persisted for at least 500 kyr. Whereas bottom water $\Delta[\text{CO}_3^{2-}]$ was near or above saturation ($\Delta[\text{CO}_3^{2-}] = 0$) prior to 950 ka, $\Delta[\text{CO}_3^{2-}]$ values after 950 ka were consistently negative in both glacials and interglacials, and they equaled modern $\Delta[\text{CO}_3^{2-}]$ of -2 $\mu\text{mol/kg}$ during interglacials MIS 19, 17, and 13 (Fig. 2c). Similarly, reconstructed $[\text{PO}_4^{3-}]$ was significantly higher after 950 ka than before, and approached modern $[\text{PO}_4^{3-}]$ of 1.6 $\mu\text{mol/kg}$ during MIS 19, 17, and 13 (Fig. 2d).

The B/Ca and Cd/Ca patterns displayed by our Site 1267 records are similar in trend but more pronounced than observations at Site 607 in the North Atlantic^{13,25} (Fig. 2c, d). Site 607 $\Delta[\text{CO}_3^{2-}]$ is higher and $[\text{PO}_4^{3-}]$ lower than at Site 1267 for the length of

the records (Fig. 2), consistent with continuously better ventilation in the deep North Atlantic compared to the abyssal Southeast Atlantic between 1,400 and 400 ka.

Quantifying deep Atlantic DIC increase

Synchronous but opposing $\Delta[\text{CO}_3^{2-}]$ and $[\text{PO}_4^{3-}]$ reconstructions across the MPT (Fig. 2c-d) suggest the deep Atlantic acquired additional respired carbon after 950 ka. Whereas $[\text{PO}_4^{3-}]$ proportionally increases with DIC through the stoichiometric ratio of carbon to phosphorous in organic matter²⁶, the CO_2 released during respiration decreases $[\text{CO}_3^{2-}]$, lowering $\Delta[\text{CO}_3^{2-}]$ (Supplementary Fig. 2). Using the C:P ratio of organic matter exported to the deep ocean ($117 \pm 14:1$, ref. 26), and making simplifying assumptions about changes in Cd inventory and seawater Cd:P ratios (Supplementary Information), we quantify deep Atlantic DIC change associated with the MOC shift around 950 ka (MIS 25) in glacial and interglacial climates (Methods). In addition, we also estimate DIC change from $\Delta[\text{CO}_3^{2-}]$ by assuming constant local alkalinity (following ref. 27, Methods).

At South Atlantic Site 1267, glacial DIC after 950 ka increased by 50 ± 16 $\mu\text{mol/kg}$ reconstructed from $\Delta[\text{CO}_3^{2-}]$, and 49 ± 24 $\mu\text{mol/kg}$ from $\Delta[\text{PO}_4^{3-}]$, compared to glacial before 950 ka (Fig. 3a, b). At North Atlantic Site 607, glacial DIC similarly increased by 50 ± 30 $\mu\text{mol/kg}$ reconstructed from $\Delta[\text{CO}_3^{2-}]$, and 20 ± 18 $\mu\text{mol/kg}$ from $\Delta[\text{PO}_4^{3-}]$ (all uncertainties 1 σ). Reconstructed DIC also increased during interglacials after 950 ka; at Site 1267 by 44 ± 24 $\mu\text{mol/kg}$ from $\Delta[\text{CO}_3^{2-}]$ and by 42 ± 16 $\mu\text{mol/kg}$ from $\Delta[\text{PO}_4^{3-}]$; at Site 607 by 23 ± 23 $\mu\text{mol/kg}$ from $\Delta[\text{CO}_3^{2-}]$ and by 10 ± 18 $\mu\text{mol/kg}$ from $\Delta[\text{PO}_4^{3-}]$ (Fig. 3). The averaged deep Atlantic glacial DIC increase after 950 ka (42 ± 15

$\mu\text{mol/kg}$) agrees with the $42 \pm 22 \mu\text{mol/kg}$ DIC increase reconstructed from benthic foraminiferal B/Ca between MIS 5a and 4 prior to the LGM²⁷. The generally larger DIC increases at Site 1267 (Fig. 3a, b) likely reflect the southerly and deeper position of Site 1267 relative to Site 607, which makes Site 1267 a more sensitive indicator of the relative contributions of northern and southern sourced deep waters. Nevertheless, these similar DIC estimates from two independent proxies at separate locations lend confidence for a persistent deep Atlantic DIC increase after 950 ka that encompassed both glacial and interglacial climates.

Increased DIC after 950 ka is likely the consequence of a more aged deep-water body in the deep Atlantic, with accumulated respired CO_2 and nutrients. Nd isotope ratios show that this aged deep water is associated with reduced NADW. Glacial ϵ_{Nd} values became more positive after 950 ka at Sites 1088, 1090 and 1267 (Fig. 3c, Supplementary Fig. 1), indicating reduced NADW and/or enhanced AABW. Interglacial ϵ_{Nd} values also show more positive values after 950 ka compared with before MIS 23 at Site 1267, indicating that reduced MOC after 950 ka was not limited to glacial climates (Methods). Covariance between ϵ_{Nd} , $\Delta[\text{CO}_3^{2-}]$ and $[\text{PO}_4^{3-}]$ at Site 1267 confirms that MOC and bottom water chemistry shifts are linked (Supplementary Fig. 3).

Implications for atmospheric pCO_2

As the Southern Ocean is the principal gateway for CO_2 to escape from the deep Atlantic into the atmosphere, changes in surface Southern Ocean properties that affect CO_2 leakage likely contribute to atmospheric CO_2 variations over Pleistocene glacial cycles^{28,29}. These processes have also been invoked for the MPT³⁰, with recent studies

proposing that expanded Southern Ocean sea-ice extent¹³ or increased iron availability¹¹ lowered pCO₂ across the MPT. Upon comparing the temporal relationships of the evidence, we propose that MPT pCO₂ reduction was closely related with reduced MOC (Fig. 3). Whereas glacial Subantarctic iron deposition gradually increased between 1,400 and 1,100 ka³¹ (Fig. 3f), ice core and proxy reconstructions suggest glacial pCO₂ did not decrease before 1,000 ka^{11,15,16} (Fig. 3d). Instead, glacial pCO₂ most likely decreased at MIS 24-22, coeval with increased deep Atlantic DIC (Fig. 3a-b), lowered sea-level at MIS 22^{10,17} (Fig. 3g), and global surface ocean cooling^{5,6}. This implies that reduced pCO₂ conforms to the more extensive glacial ice sheets of the 100-kyr world. Our prediction that glacial pCO₂ was primarily reduced during MIS 24 and 22 agrees with pCO₂ simulations^{32,33}, and is testable with more detailed pCO₂ reconstructions from this critical interval.

Lower pCO₂ is plausibly linked to increased deep Atlantic DIC by more complete Southern Ocean nutrient utilisation and/or increased deep ocean stratification^{29,34,35} after 950 ka. As indicated above, we do not believe that iron fertilisation primarily drove lower pCO₂ after 950 ka. Instead, the coherent timing with MOC suggests a circulation-driven process. The Southern Ocean's residual overturning circulation is driven by surface density fluxes³⁶, and it has been proposed that changes to these density fluxes in glacial climates are dynamically linked to deep ocean stratification^{35,37} and pCO₂³⁸. Specifically, reduced rates of surface water densification combined with a northward shift of Southern Ocean upwelling during glacial climates³⁸ increases the surface ocean residence time of nutrient-rich waters, enhancing nutrient utilisation and CO₂ uptake before new AABW is formed³⁸. Both stratification and enhanced nutrient utilisation would increase deep ocean

DIC by preferentially segregating regenerated products (nutrients and carbon) in the deep ocean. Deep ocean $\Delta[\text{CO}_3^{2-}]$ consequently decreases, driving a transient carbonate (CaCO_3) dissolution event³⁹ that may increase alkalinity and foster additional ocean CO_2 storage^{40,41}, although the mechanisms and timescales of such alkalinity feedbacks are debated²⁵. Accounting for an alkalinity increase slightly elevates our estimate of deep Atlantic DIC increase from B/Ca, but does not fundamentally change our results (Methods).

The above conceptual arguments are supported by similar magnitudes of pCO_2 reduction at the MPT and in LGM model simulations of reduced MOC coupled to reduced Southern Ocean CO_2 outgassing. Assuming our pooled estimate of glacial DIC increase ($42 \pm 15 \text{ } \mu\text{mol/kg}$) characterised the entire deep Atlantic below 3 km, then its carbon inventory increased by $51 \pm 18 \text{ Gt C}$ during glaciations after 950 ka (Methods). Quantifying pCO_2 reduction from this DIC increase requires simulating the new equilibrium reached between the ocean and atmospheric C reservoirs following the DIC increase. Model simulations of the LGM provide a comparable constraint where the effects of MOC and reduced Southern Ocean CO_2 outgassing on deep Atlantic DIC and pCO_2 are quantified. In these simulations, combining reduced MOC with reduced Southern Ocean CO_2 outgassing increases deep ocean DIC through a stronger biological pump^{27,29,43} and lowers pCO_2 by 13 to 45 ppm^{29,42,43}. The consistency between these pCO_2 estimates and the reconstructed 20-40 ppm glacial pCO_2 decrease across the MPT^{11,15,16} (Fig. 3e) supports a MOC and Southern Ocean pathway for MPT glacial pCO_2 reduction. This implies that the ‘ventilation volume’ CO_2 sequestration hypothesis

for the LGM^{29,43} first initiated with the major glacial MOC reduction after 950 ka¹², and drove lower glacial pCO₂ levels during ensuing 100-kyr glacial cycles (Fig. 4).

Our averaged estimate for interglacial DIC increase (30±16 μmol/kg) implies a deep Atlantic carbon inventory increase of 36±20 Gt C during interglacials after 950 ka. This evidence for greater interglacial deep ocean carbon storage after 950 ka may have contributed to the ~30 to 40 ppm lower pCO₂ during the “lukewarm” interglacials between 450 and 900 ka^{15,44}, although identifying mechanisms for this requires further study.

Perspective

An enduring mystery surrounding the MPT is how the cryosphere, ocean, and carbon cycle interacted to initiate 100-kyr climate cycles at this particular time in Earth history. While erosion of Northern Hemisphere regolith has been a central hypothesis for the MPT^{5,7,9,11}, we suggest that an expanded Antarctic Ice Sheet played a key role in unleashing the ocean carbon cycle feedbacks necessary for 100-kyr cycles (Fig. 4). Atmospheric cooling associated with growth of the East Antarctic Ice Sheet to marine-based margins around 1000 ka^{8,45} (Fig. 3e) would have enhanced the production and export of Antarctic sea-ice³⁷, expanding the spatial extent^{35,37} but reducing the intensity of surface water densification around Antarctica³⁸. This increased residence time of nutrient-rich waters would have enhanced nutrient utilisation and therefore CO₂ uptake. Increased Antarctic surface “buoyancy loss” may also have directly facilitated expanded AABW and enhanced deep ocean stratification^{35,37}, although we cannot rule out Northern Hemisphere forcing of the MOC shift through reduced NADW production (Fig. 4c).

Future studies of Southern Ocean nutrient utilisation and deep ocean stratification across the MPT can elucidate how these two mechanisms contributed to MPT pCO₂ decline.

In summary, our results demonstrate that the rise of 100-kyr ice age cycles is intimately linked with increased deep ocean carbon storage associated with weakened MOC. The key roles for ocean circulation and the carbon cycle in the natural climate transition at the MPT raise critical questions on the future behaviour of these climate components under anthropogenic warming. By the end of the century, Atlantic MOC is projected to decrease by potentially similar magnitude (~20%) as the MOC reduction at the MPT, albeit with high uncertainty^{46,47}. While we have demonstrated that a weaker MOC related to increased deep ocean carbon storage at the MPT, future deep ocean carbon sequestration depends on how future MOC interacts with the evolution of the biological pump and deep ocean stratification. In contrast to the MPT, projected future reductions of Southern Ocean surface density^{37,48} and reduced anthropogenic CO₂ uptake by NADW under MOC weakening⁴⁹ may diminish the future carbon sequestration potential of the deep Atlantic Ocean.

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Figure Captions.

Figure 1. Location of sediment cores and modern Atlantic Ocean hydrography. a) Site 1267²⁰ (cyan star) is located in the South Atlantic, along with cores comprising the SUSAS stack⁵⁰ (gray circles), and Sites 1088 and 1090^{12,14,31}. Site 607^{13,25} (orange diamond) is located in the North Atlantic. Colours denote surface ocean potential density from World Ocean Atlas 2013 data (www.nodc.noaa.gov/OC5/woa13/); white X's denote regions of modern-day deep-water formation. b) Meridional cross-section showing modern-day phosphate (colour) and salinity (contours indicate isolines) (composite of CLIVAR A20 and A13.5, www.clivar.org). Phosphate-poor NADW fills the Angola Basin to >4 km depth, while phosphate-rich AABW inflow is limited by Walvis Ridge. Phosphate-rich Antarctic Intermediate Water (AAIW) extends northward above 2 km depth. Plots created in Ocean Data View (<http://odv.awi.de>).

Figure 2. Site 1267 and 607 circulation and trace element records across the MPT. a) Site 1267 $\delta^{18}\text{O}_b$ ²¹ (black) and $\delta^{13}\text{C}_b$ ^{21,22} (brown); red and blue numbers denote interglacial and glacial stages, respectively. b) Site 1267 ϵ_{Nd} ; opposite axis gives corresponding percentage North Atlantic sourced water (NAW, calculation in Methods). c) B/Ca from 607^{13,25} (gray circles) and 1267 (filled squares); opposite axis gives $\Delta[\text{CO}_3^{2-}]$ (Methods). d) Cd/Ca from 607¹³ (gray circles) and 1267 (outlined squares); opposite axis gives $[\text{PO}_4^{3-}]$ (Methods). Error bars are replicate 1s, and are greater than long-term internal precision (Methods). Red, blue and gray squares indicate Site 1267 interglacial, glacial and transitional samples, respectively. Horizontal lines in c) and d)

indicate modern $\Delta[\text{CO}_3^{2-}]$ and $[\text{PO}_4^{3-}]$ at 1267 (blue) and 607 (orange). Vertical gray shading highlights MIS 25 through 22.

Figure 3. Changes in ocean CO₂ feedbacks and climate across the MPT. ΔDIC reconstructed from $\Delta[\text{CO}_3^{2-}]$ (a) and $\Delta[\text{PO}_4^{3-}]$ (b) at Sites 1267 (filled squares) and 607 (open squares). c) South Atlantic MOC from ϵ_{Nd} at Sites 1267 (squares) and 1088¹² (circles). Red/blue symbols in a-c indicate interglacials/glacials, respectively. d) pCO₂ from EPICA Dome C⁴⁴ (black line), Allan Hills¹⁶ (black bar at 1,000 ka), planktic foraminiferal boron isotopes (red/blue/gray circles are interglacials/glacials/transitional from ref. 15; cyan curve, ref. 11), and modeled³³. Yellow shading indicates average pCO₂ for 0-420, 420-900, and 950-2,000 ka. e) Terrigenous mass accumulation rate from Site 745 indicating East Antarctic Ice Sheet activity⁴⁵. f) Iron mass accumulation rate in the Subantarctic Atlantic (Site 1090)³¹. g) Eustatic sea level from South Pacific $\delta^{18}\text{O}_{\text{sw}}$ ¹⁰ (cyan line) and Mediterranean planktic foraminifer $\delta^{18}\text{O}$ ¹⁷ (yellow points).

Figure 4. Schematics of proposed coupling between cryosphere, Atlantic MOC, and carbon cycle responses across the MPT. Glacial Atlantic MOC cross-sections before and after 950 ka are in a) and c), respectively. b) and d) show Antarctic processes within dashed regions of a) and c). Curved arrows trace AABW and NADW pathways; arrow colour indicates $[\text{PO}_4^{3-}]$ (Fig. 1) and thickness indicates deep-water contribution (thicker is greater contribution). RFZ= Romanche Fracture Zone. Red arrows indicate ocean CO₂ release; orange dots indicate dust deposition. Black curved lines in b) and d) indicate brine rejection. EAIS= East Antarctic Ice Sheet. Before 950 ka, a smaller EAIS limits Antarctic sea-ice coverage, shortening the path for Antarctic surface waters before subduction, giving a larger fraction of unutilised nutrients in AABW and greater CO₂ release (b). NADW dominates Atlantic MOC (a). After 950 ka, EAIS expands to marine-based margins and sea-ice expands. Antarctic surface waters upwell farther north, travelling further before subduction, decreasing $[\text{PO}_4^{3-}]$ in AABW and lowering CO₂ outgassing (d). AABW dominates deep Atlantic circulation, effectively trapping CO₂ (cross-hatching, c).

Methods

ODP Site 1267 and sample selection. ODP Site 1267 (28°5.89' S, 1°42.66' E, 4.36 km water depth) was cored near the northwestern margin of Walvis Ridge in the Angola Basin²⁰. Site 1267 chronology was established by aligning *Cibicidoides wuellerstorfi* $\delta^{18}\text{O}^{21}$ to the LR04 global $\delta^{18}\text{O}_b$ stack⁵¹. For this study, *C. wuellerstorfi* specimens >250 μm were picked from samples reflecting glacial-interglacial extremes (indicated by $\delta^{18}\text{O}_b$) spanning MIS 46 to MIS 12, and every 25 cm (every ~3 kyr) between MIS 30 and MIS 22 (Supplementary Fig. 4). Our records do not extend beyond MIS 12 due to limited *C. wuellerstorfi* in younger sediments. Glacial and interglacial extremes in B/Ca and Cd/Ca records from Site 607^{13,25} were also selected based on $\delta^{18}\text{O}_b$ (Supplementary Fig. 4).

Trace element analyses. For Site 1267, six to twelve *C. wuellerstorfi* specimens were crushed between two methanol-cleaned glass slides and transferred to HCl-cleaned 0.5 mL polypropylene vials. When necessary, adjacent samples were combined to provide 200 to 300 μg of calcite. Crushed samples were cleaned via clay removal in methanol and boron-free MilliQ water, removal of Fe-Mn oxides and authigenic Mn carbonates by reduction with an ammonium-citric acid-buffered hydrazine ($\text{NH}_4(\text{OH})\text{-C}_6\text{H}_8\text{O}_7\text{-N}_2\text{H}_4$) solution, organic oxidation with NaOH-buffered H_2O_2 , and a weak acid leach with 0.001 N HNO_3 to remove any adsorbed contaminants, following standard protocols⁵²⁻⁵⁴. Samples were transferred to HCl-cleaned 2 mL polypropylene vials after the oxidation step. Initial tests showed elevated Mn and Fe concentrations (> 100 $\mu\text{mol/mol}$) without reductive cleaning, and thus reductive cleaning was performed on all samples. Removal of Fe and Mn phases is critical for reducing Cd blank⁵³ and does not affect B/Ca ratios⁵⁵. All sample preparation was performed in a boron-free HEPA laminar flow bench at the Lamont-Doherty Earth Observatory (LDEO).

Cleaned samples were dissolved in 0.8 to 1.6 mL of ultrapure 2% HNO_3 immediately prior to analysis, with the volume of acid adjusted by sample mass to obtain ~50 ppm [Ca^{2+}] solutions. Trace element intensities were measured using a Thermo Scientific iCAP Q inductively coupled plasma mass spectrometer at LDEO⁵⁶ following methods adapted from ref. 57. Low mass element ^{11}B was measured in standard mode;

²³Na, ²⁷Al, ⁴³Ca, ⁵⁵Mn, ⁵⁶Fe, and ¹¹¹Cd were measured in kinetic energy discrimination mode using a He collision cell to minimise polyatomic interferences. To minimise matrix effects, element intensities were corrected with an internal multi-element quality control standard measured every ten samples. Concentrations and element ratios (relative to Ca) were calculated from element intensities measured on a multi-element stock solution prepared gravimetrically from trace element-grade stock solutions. This solution was diluted to [Ca²⁺] of 5 to 125 ppm and each dilution was then measured, creating calibrations of element intensity to concentration. External precision was evaluated using nine full procedural replicates from Site 1267, with 1s pooled replicate precision of ±6.6 µmol/mol for B/Ca and ±0.009 µmol/mol for Cd/Ca. Long-term internal precision (1s) of an in-house quality control standard was ±2.9 µmol/mol for B/Ca and ±0.006 µmol/mol for Cd/Ca.

Site 607 B/Ca and Cd/Ca ratios were previously measured via similar protocols^{13,25}. Pooled replicate precision (1s) for Site 607 B/Ca and Cd/Ca was ±5.9 and ±0.007 µmol/mol, respectively¹³. No attempt was made to correct for laboratory B/Ca or Cd/Ca offsets. If laboratory offsets existed, they would not change the results of our study because each core was measured in separate laboratories and we independently calculate ΔDIC from B/Ca and Cd/Ca variations within each core (see below).

Neodymium isotope analyses. Nd isotopes (¹⁴³Nd/¹⁴⁴Nd) were measured on Fe-Mn encrusted planktic foraminifera^{12,58}. Between 20 and 30 mg of planktic foraminifera from the >300 µm size fraction were crushed, ultrasonicated and resuspended in MilliQ water and methanol to remove detrital contaminants. Each sample was treated individually to ensure that sufficient rinsing steps were applied. Cleaned fragments were dissolved in dilute acetic acid, and the resulting solution centrifuged. The supernatant was transferred to clean PFA beakers and Nd was separated using Eichrom Tru-Spec and Ln-Spec resins. Nd isotopes were determined by multi-collector inductively coupled mass spectrometry on a ThermoScientific Neptune-Plus at LDEO. Instrumental drift was corrected by sample-standard bracketing (SSB) using JNdi-1 as the primary standard⁵⁹, with matching standard and sample Nd concentrations. Nd isotopes are expressed as ε_{Nd} values, which give deviations in parts per 10⁴ from the average ¹⁴³Nd/¹⁴⁴Nd value of chondritic

meteorites (0.512638)⁶⁰. External analytical reproducibility ($\pm 2s$) was 0.28 and 0.43 ϵNd units (± 0.000014 and 0.000021 absolute) for the two analytical sessions where Site 1267 samples were measured.

Converting B/Ca, Cd/Ca to DIC_{resp}. Global coretop calibration shows that *C. wuellerstorfi* B/Ca records $\Delta[CO_3^{2-}]^{23,61-63}$, where $\Delta[CO_3^{2-}] = [CO_3^{2-}]_{in situ} - [CO_3^{2-}]_{saturation}$. Site 607 and 1267 B/Ca was converted to $\Delta[CO_3^{2-}]$ using B/Ca ($\mu mol/mol$) = $1.14 \pm 0.04^*$ $\Delta[CO_3^{2-}]$ ($\mu mol/kg$) + 176.6 ± 1.0 (ref. 63).

Both $[CO_3^{2-}]_{in situ}$ and $[CO_3^{2-}]_{saturation}$ are relatively insensitive to changes in temperature, salinity, and pressure. For instance, projected changes in deep ocean temperature, salinity and pressure on glacial-interglacial timescales negligibly influence $\Delta[CO_3^{2-}]$ compared to instrumental B/Ca precision⁶⁴. Therefore, downcore changes in $\Delta[CO_3^{2-}]$ primarily reflect changes in deep ocean $[CO_3^{2-}]_{in situ}$ driven by changes in the difference between total alkalinity (TA) and dissolved inorganic carbon (DIC) (Supplementary Fig. 2; ref. 27). If the quantity TA-DIC increases, the greater excess of TA relative to DIC implies a shift in DIC speciation toward CO_3^{2-} and thus increased $[CO_3^{2-}]$. Conversely, a decrease in the quantity TA-DIC implies DIC speciation shifts toward CO_2 , and thus $[CO_3^{2-}]$ decreases.

Changes in $\Delta[CO_3^{2-}]$ at Sites 1267 and 607 are a function of TA and DIC (Supplementary Fig. 2):

$$(\Delta[CO_3^{2-}]_{measured} - \Delta[CO_3^{2-}]_{reference}) = 0.54 * (\Delta TA - \Delta DIC) \quad (1)$$

where $\Delta[CO_3^{2-}]_{measured} - \Delta[CO_3^{2-}]_{reference}$ gives the $\Delta[CO_3^{2-}]$ difference from a reference $\Delta[CO_3^{2-}]$ value (here, the average pre-950 ka interglacial $\Delta[CO_3^{2-}]$ of $+14.8 \mu mol/kg$ at Site 1267 and $+39.7 \mu mol/kg$ at Site 607), and ΔTA and ΔDIC are the differences in TA and DIC between each sample relative to average pre-950 ka interglacial TA and DIC. Note that while we use $\Delta[CO_3^{2-}]$ instead of $[CO_3^{2-}]_{in situ}$ (as used by ref. 27 and 52), Equation 1 is identical using either parameter because $[CO_3^{2-}]_{saturation}$ is assumed constant due to the small influence of temperature, salinity, and pressure variations on $[CO_3^{2-}]_{saturation}$ (see above). The slope of Equation 1 ($k=0.54$) is less than one due to a larger

contribution from non-carbonate species (particularly borate anion) to TA as the quantity TA-DIC increases. We calculated k by varying TA and DIC at the present-day bottom-water temperature (2.4°C), salinity (34.89) and pressure (4429 dbar) of Site 1267 in CO2SYS (ref. 65). Following this for Site 607 also returns k of 0.54. In contrast, Yu *et al.*²⁷ used $k=0.59$ from regression of oceanographic data below 2.5 km depth in the Atlantic Ocean, but this approach includes influences from pressure, temperature, and salinity variations within the dataset used for their regression. Our slope is more appropriate because the expected temperature, salinity and pressure variations at Sites 1267 and 607 across the late Pleistocene are smaller than the present-day ranges of these parameters in the Atlantic Ocean below 2.5 km depth.

Two processes primarily control deep ocean TA and DIC: organic matter respiration and calcium carbonate (CaCO₃) dissolution (Supplementary Fig. 2). Explicitly accounting for the effects of respiration and dissolution in Eq. 1 and substituting pre-950 ka interglacial $\Delta[\text{CO}_3^{2-}]$ for $\Delta[\text{CO}_3^{2-}]_{\text{reference}}$ gives:

$$(\Delta[\text{CO}_3^{2-}]_{\text{measured}} - 14.8) = 0.54 * ((\Delta\text{TA} - \Delta\text{DIC})_{\text{resp}} + (\Delta\text{TA} - \Delta\text{DIC})_{\text{diss}}) \quad (2)$$

for Site 1267. (For Site 607, 39.7 $\mu\text{mol/kg}$ is subtracted from $\Delta[\text{CO}_3^{2-}]_{\text{measured}}$ to normalize to average pre-950 ka interglacial $\Delta[\text{CO}_3^{2-}]$). Here *resp* and *diss* denote changes from organic matter respiration and dissolution, respectively. Respiring 1 μmol of organic matter in 1 kg seawater increases DIC by 1 $\mu\text{mol/kg}$ and decreases TA by 0.14 $\mu\text{mol/kg}$ (ref. 26). This decreases the quantity TA-DIC by 1.14 $\mu\text{mol/kg}$ and thus decreases $[\text{CO}_3^{2-}]_{\text{in situ}}$ by 0.62 $\mu\text{mol/kg}$. Dissolution of 1 μmol of CaCO₃ in 1 kg seawater increases DIC by 1 $\mu\text{mol/kg}$ and increases TA by 2 $\mu\text{mol/kg}$, which increases TA-DIC by 1 $\mu\text{mol/kg}$ and thus raises $[\text{CO}_3^{2-}]_{\text{in situ}}$ by 0.54 $\mu\text{mol/kg}$. Using these stoichiometric ratios of TA- and DIC-change during respiration and dissolution, Eq. 2 becomes:

$$(\Delta[\text{CO}_3^{2-}]_{\text{measured}} - 14.8)(\mu\text{mol} / \text{kg}) = 0.54 * (0.5\Delta\text{TA}_{\text{diss}} - 1.14\Delta\text{DIC}_{\text{resp}}) \quad (3)$$

for Site 1267. As in Eq. 1, $\Delta\text{TA}_{\text{diss}}$ and $\Delta\text{TA}_{\text{resp}}$ values are calculated relative to pre-950 ka interglacial averages.

Equation 3 has three unknowns: $\Delta[\text{CO}_3^{2-}]$, $\Delta\text{TA}_{\text{diss}}$, and $\Delta\text{DIC}_{\text{resp}}$. B/Ca constrains $\Delta[\text{CO}_3^{2-}]$, but one more constraint is required to solve Eq. 3. One approach is to assume constant deep ocean alkalinity²⁷ ($\Delta\text{TA}_{\text{diss}}=0$), which allows Eq. 3 to be solved for $\Delta\text{DIC}_{\text{resp}}$. Previous applications of B/Ca to reconstruct DIC assumed constant TA over short study intervals^{27,66} (<50 kyr). In contrast, our 1-Myr record covers a climate transition associated with notable changes in deep-sea carbonate sedimentation^{25,49,67} (Supplementary Fig. 5). Thus, it may not be practical to assume that $\Delta\text{TA}_{\text{diss}}=0$ (see below).

Lacking established paleo-proxies for TA, here we also calculate $\Delta\text{DIC}_{\text{resp}}$ using *C. wuellerstorfi* Cd/Ca, an independent nutrient proxy that is unlikely affected by $\Delta\text{TA}_{\text{diss}}$. *Cibicidoides wuellerstorfi* Cd/Ca relates to deep ocean [Cd] through the following relationship:

$$[\text{Cd}]_{\text{sw}} (\text{nmol} / \text{kg}) = \frac{10 * (\text{Cd} / \text{Ca})_{\text{C.wuellerstorfi}}}{2.9} \quad (4)$$

where 2.9 is the partition coefficient of Cd into calcitic foraminifera below 3 km depth and 10 is the seawater $[\text{Ca}^{2+}]$ in mmol/kg^{24,68,69}. $[\text{Cd}]_{\text{sw}}$ relates to $[\text{DIC}]_{\text{resp}}$ through stoichiometric relationships in seawater and organic matter. In the modern Atlantic Ocean, $[\text{Cd}]_{\text{sw}}$ is strongly, but nonlinearly, correlated with phosphate concentration ($[\text{PO}_4^{3-}]$)⁷⁰⁻⁷³. This nonlinearity might reflect Cd uptake by phytoplankton⁷¹ possibly modulated by the local abundance of other trace elements⁷². Recent high-quality $[\text{Cd}]_{\text{sw}}$ measurements taken during the GEOTRACES program in the Atlantic Ocean⁷²⁻⁷⁵ exhibit a strong nonlinear correlation to $[\text{PO}_4^{3-}]$, which is linearized by using the square root of $[\text{Cd}]_{\text{sw}}$ (Supplementary Fig. 2c):

$$[\text{PO}_4^{3-}] (\mu\text{mol} / \text{kg}) = \frac{\sqrt{[\text{Cd}]_{\text{sw}}} - 0.062}{0.361} \quad (5)$$

Cibicidoides wuellerstorfi Cd/Ca is used to calculate $\Delta[\text{Cd}]_{\text{sw}}$ using Equation 4, and then $\Delta\text{DIC}_{\text{resp}}$ associated with $\Delta[\text{Cd}]_{\text{sw}}$ is calculated assuming a Redfield C:P ratio of 117±14:1 (ref. 26):

$$\Delta DIC_{resp} (\mu mol / kg) = (117 \pm 14) * \frac{\sqrt{\Delta [Cd]_{sw}} - 0.062}{0.361} \quad (6)$$

As in Equation 3, ΔDIC_{resp} and $\Delta [Cd]_{sw}$ are calculated as differences relative to pre-950 ka interglacial averages.

Reconstructed ΔDIC_{resp} from Cd/Ca (Equation 6) largely agrees with ΔDIC_{resp} calculated from B/Ca (Equation 3) assuming $\Delta TA_{diss}=0$ over our study interval at Site 1267 (Supplementary Fig. 6). This correspondence argues against major variations in uncertainty sources that affect one—but not both—proxies. Specifically, this correspondence suggests that any changes in ΔTA_{diss} (influencing B/Ca) are smaller than the (relatively large) uncertainty on B/Ca-derived ΔDIC_{resp} (discussed below). Similarly, this argues against secondary influences on Cd/Ca-derived ΔDIC_{resp} estimates (Supplementary Information).

Quantifying deep Atlantic carbon storage. Following ref. 27, we quantify deep Atlantic carbon storage as follows:

$$\Delta C(Gt) = \frac{-12.011 * \Delta DIC_{Atlantic>3km} * m_{Atlantic>3km}}{10^{21}} \quad (7)$$

where $\Delta DIC_{Atlantic>3km}$ is the pooled average DIC increase at Sites 1267 and 607 in $\mu mol/kg$ ($42 \pm 15 \mu mol/kg$ for glacial and $30 \pm 16 \mu mol/kg$ for interglacials), $m_{Atlantic>3km}$ is the mass of water in the Atlantic Ocean below 3 km depth ($1.01 * 10^{20}$ kg), 12.011 is the mass of carbon (g/mol), and factor 10^{21} converts units from μg C to Gt C⁷⁶. This calculation assumes that the magnitude of DIC increase reconstructed from Sites 1267 and 607 characterises the entire Atlantic Ocean below 3 km.

Significance Testing. Statistical comparisons of mean glacial/interglacial ϵ_{Nd} and reconstructed DIC before and after 950 ka was performed using one-tailed, two-sample t -test with unequal variance (Welch's t -test) in MATLAB (function `ttest2`). Significance was evaluated with $\alpha=0.05$. Mean values and p -values for statistical tests are given in Supplementary Table 1.

Influence of alkalinity on pCO₂. Records of deep ocean sedimentary CaCO₃ content suggest pronounced changes in CaCO₃ sedimentation at the MPT. Expanded AABW resulting from reduced MOC lowered CaCO₃ preservation throughout the deep Atlantic at the MPT, as evidenced by carbonate %²⁵, sediment lightness (L*) and magnetic susceptibility records⁵⁰ (Supplementary Fig. 5). However, Site 1267 and 607 $\Delta[\text{CO}_3^{2-}]$ reconstructions do not show clear evidence for a transient $\Delta[\text{CO}_3^{2-}]$ rise across the MOC reduction (Fig. 2), as might be expected to occur within 5-10 kyr from “classic” carbonate compensation³⁹. Site 1267 has an average sampling resolution of 9.2 kyr between 1,000 and 875 ka, and it is possible that this record lacks sufficient resolution to reconstruct a rapid carbonate compensation event. Site 607 has an average resolution of ~3 kyr, yet also does not show transient $\Delta[\text{CO}_3^{2-}]$ rises²⁵.

One possibility is that any $\Delta[\text{CO}_3^{2-}]$ rise from carbonate compensation is masked within these records; i.e., $\Delta[\text{CO}_3^{2-}]$ decreases less than expected from DIC increase alone due to the counteracting influence of increased TA from regional carbonate dissolution. In this scenario, ΔDIC derived from B/Ca would systematically underestimate true ΔDIC . To quantify this underestimation, Equation 3 shows that a 1 $\mu\text{mol/kg}$ increase in TA from CaCO₃ dissolution increases $\Delta[\text{CO}_3^{2-}]$ by 0.27 $\mu\text{mol/kg}$. Estimates of the TA contribution from CaCO₃ dissolution in the modern ocean (termed TA*, ref. 77) show TA* of 0-10 $\mu\text{mol/kg}$ for NADW and 20-30 $\mu\text{mol/kg}$ for AABW. Assuming that the MOC shift at Site 1267 was associated with a 30 $\mu\text{mol/kg}$ TA increase from enhanced CaCO₃ dissolution, $\Delta[\text{CO}_3^{2-}]$ would increase by 8.1 $\mu\text{mol/kg}$. This increase equates to ΔDIC of -13 $\mu\text{mol/kg}$, giving a revised post-950 ka glacial ΔDIC estimate of $63 \pm 16 \mu\text{mol/kg}$ from $\Delta[\text{CO}_3^{2-}]$ from Sites 1267 and 607, which is indistinguishable from the $\Delta[\text{PO}_4^{3-}]$ -derived ΔDIC estimate of $49 \pm 24 \mu\text{mol/kg}$.

Based on Site 607 $\Delta[\text{CO}_3^{2-}]$, Sosdian et al.²⁵ present a new model of carbonate compensation driven by changes in regional carbonate burial, and not local dissolution. In this model, the relationship between deep Atlantic TA and global ocean TA depends on Pacific CaCO₃ burial, giving a more complex relationship between deep Atlantic TA and potential CO₂ drawdown. Regardless, while the precise magnitude of ΔDIC may be sensitive to TA change from carbonate compensation and compensation mechanisms, the

conclusion of increased DIC in the deep Atlantic coeval with reduced MOC is not contingent on assuming constant TA.

Mixing between NADW and AABW. Supplementary Figure 3 shows relationships between Site 1267 $[\text{PO}_4^{3-}]$, $[\text{CO}_3^{2-}]$, $\delta^{13}\text{C}^{22}$, and relative NADW/AABW contributions from ϵ_{Nd} . Mixing curves are calculated using modern day endmember values for NADW and AABW defined from hydrographic data for ϵ_{Nd} , $[\text{PO}_4^{3-}]$ and $[\text{CO}_3^{2-}]$, and using late Holocene benthic foraminifer $\delta^{13}\text{C}$ for $\delta^{13}\text{C}_{\text{DIC}}$. Foraminifera were chosen to define the $\delta^{13}\text{C}_{\text{DIC}}$ endmembers considering the impact of anthropogenic carbon on hydrographic $\delta^{13}\text{C}_{\text{DIC}}$ (the ‘Suess effect’), particularly in NADW⁷⁸. Endmember values for ϵ_{Nd} and [Nd]: NADW = -14.0 ± 0.5 ϵ units and [Nd] = 23.6 pmol/kg, AABW = -8.0 ± 0.5 ϵ units and [Nd] = 26.1 pmol/kg (Y. Wu, pers. comm.); $[\text{PO}_4^{3-}]$: NADW = 1.1 ± 0.05 $\mu\text{mol/kg}$, AABW = 2.25 ± 0.01 $\mu\text{mol/kg}$ ⁷⁴; $[\text{CO}_3^{2-}]$: NADW = 120 ± 10 $\mu\text{mol/kg}$, AABW = 85 ± 5 $\mu\text{mol/kg}$ ⁷⁴; $\delta^{13}\text{C}$: NADW = 1.36 ± 0.08 ‰ (0-5 ka from BOFS 11K, ref. 64); AABW = 0.27 ± 0.1 ‰ (coretop from TN057-21; ref. 79). Mixing curves are calculated from simple two-component concentration and isotope mixture equations (example below for ϵ_{Nd}):

$$\epsilon_{\text{Nd}_{\text{mixture}}} = \frac{f_{\text{NADW}} * [\text{Nd}]_{\text{NADW}} * \epsilon_{\text{Nd}_{\text{NADW}}} + f_{\text{AABW}} * [\text{Nd}]_{\text{AABW}} * \epsilon_{\text{Nd}_{\text{AABW}}}}{f_{\text{NADW}} * [\text{Nd}]_{\text{NADW}} + f_{\text{AABW}} * [\text{Nd}]_{\text{AABW}}} \quad (8)$$

Preformed nutrients in AABW. In NADW and AABW endmembers, $[\text{PO}_4^{3-}]$ is composed of preformed PO_4^{3-} and regenerated PO_4^{3-} components^{80,81}:

$$[\text{PO}_4^{3-}]_{\text{total}} (\mu\text{mol} / \text{kg}) = [\text{PO}_4^{3-}]_{\text{preformed}} + [\text{PO}_4^{3-}]_{\text{regenerated}} \quad (9)$$

Following ref. 82, we use $[\text{PO}_4^{3-}]_{\text{preformed}}$ of 0.8 $\mu\text{mol/kg}$ for NADW and 1.5 $\mu\text{mol/kg}$ for AABW. These values agree with $[\text{PO}_4^{3-}]_{\text{preformed}}$ calculated from GEOTRACES⁷⁴ apparent oxygen utilisation (AOU) data ($[\text{PO}_4^{3-}]_{\text{preformed}}$ of 0.78 ± 0.03 and 1.57 ± 0.03 $\mu\text{mol/kg}$ for NADW and AABW, respectively, using AOU of 55 ± 5 and 115 ± 5 $\mu\text{mol/kg}$ for NADW and AABW, respectively, and an $\text{O}_2:\text{PO}_4^{3-}$ ratio of 170 from ref. 26; calculation in ref. 82).

Additional curves in Supplementary Figure 3 show mixing between modern NADW and AABW with $[\text{PO}_4^{3-}]_{\text{preformed}}$ reduced by 25% and 50%, giving AABW $[\text{PO}_4^{3-}]$

$[\text{PO}_4^{3-}]_{\text{preformed}}$ of 1.13 and 0.9 $\mu\text{mol/kg}$, respectively. Note that $[\text{PO}_4^{3-}]_{\text{total}}$ is kept constant, any
 $[\text{PO}_4^{3-}]_{\text{preformed}}$ decrease is added to $[\text{PO}_4^{3-}]_{\text{regenerated}}$ ⁸¹. The addition of $[\text{PO}_4^{3-}]_{\text{regenerated}}$ adds
 regenerated DIC, increasing DIC (117:1 ratio²⁶), and thus decreasing $[\text{CO}_3^{2-}]$. The effect
 of DIC addition on $[\text{CO}_3^{2-}]$ is calculated using Equation 3 (above). The effect on $\delta^{13}\text{C}$ is
 calculated by assuming the added DIC possesses a constant $\delta^{13}\text{C}$ of -25‰ corresponding
 to organic matter. Two-component mixing curves are then recalculated using the modern-
 day NADW endmember values and the updated AABW endmember $[\text{CO}_3^{2-}]$ and $\delta^{13}\text{C}$
 values.

Cross-plots of $[\text{PO}_4^{3-}]$, $[\text{CO}_3^{2-}]$ and $\delta^{13}\text{C}$ with ϵ_{Nd} show that reconstructed $[\text{PO}_4^{3-}]$
 corresponds to mixing between modern NADW and AABW endmembers, while $[\text{CO}_3^{2-}]$
 and benthic $\delta^{13}\text{C}$ imply mixing with AABW possessing lower $[\text{CO}_3^{2-}]$ and $\delta^{13}\text{C}_{\text{DIC}}$ than
 modern (Supplementary Fig. 3). These observations can be reconciled by a higher
 regenerated to preformed PO_4^{3-} ratio in mid-Pleistocene AABW compared to today. A
 shift from $[\text{PO}_4^{3-}]_{\text{preformed}}$ to $[\text{PO}_4^{3-}]_{\text{regenerated}}$ would not change total $[\text{PO}_4^{3-}]$ ⁸¹, but would
 decrease AABW $[\text{CO}_3^{2-}]$ and $\delta^{13}\text{C}$ through excess regenerated ^{13}C -depleted organic
 matter and sluggish air-sea gas exchange in the Southern Ocean. Mixing arrays calculated
 with varying regenerated to preformed PO_4^{3-} ratios suggest that AABW in the Angola
 Basin may have had up to 50% lower $[\text{PO}_4^{3-}]_{\text{preformed}}$ during the MPT compared to today,
 particularly in glacials after 950 ka (Supplementary Fig. 3). This level of decreased
 AABW $[\text{PO}_4^{3-}]_{\text{preformed}}$ is similar to estimates for the Last Glacial Maximum (LGM)⁸¹,
 when AABW $\delta^{13}\text{C}$ and $\Delta[\text{CO}_3^{2-}]$ were lower than today^{14,63}, and is consistent with the
 proposed coupling between Southern Ocean nutrient utilisation and expanded AABW
 through Southern Ocean surface buoyancy forcing³⁸.

Because modern AABW has more preformed nutrients than NADW, assuming
 that modern AABW expanded in the past would have ventilated a greater portion of the
 deep Atlantic through the Southern Ocean, where the biological pump is less efficient. In
 this scenario, reduced MOC would both increase deep Atlantic DIC and increase
 pCO_2 ^{29,80}. For the MPT, if expanded AABW possessed 50% lower $[\text{PO}_4^{3-}]_{\text{preformed}}$, the
 effect of preformed nutrients on pCO_2 would be minor. This is because AABW would
 replace NADW of similar $[\text{PO}_4^{3-}]_{\text{preformed}}$, implying that deep Atlantic $[\text{PO}_4^{3-}]_{\text{preformed}}$

stayed relatively constant. This is critical for pCO₂ because it implies that reduced MOC was not associated with a less efficient biological pump, and further implies that reduced MPT pCO₂ required coupling between MOC and CO₂ outgassing in the Southern Ocean (Fig. 4).

Data availability. The datasets generated during the current study are available as Supplementary Tables 2 through 5 in the Supplementary Data, and will be publicly accessible via the National Centers for Environmental Information (NCEI) following acceptance. A persistent web link to the dataset will be provided at that time.

Methods References.

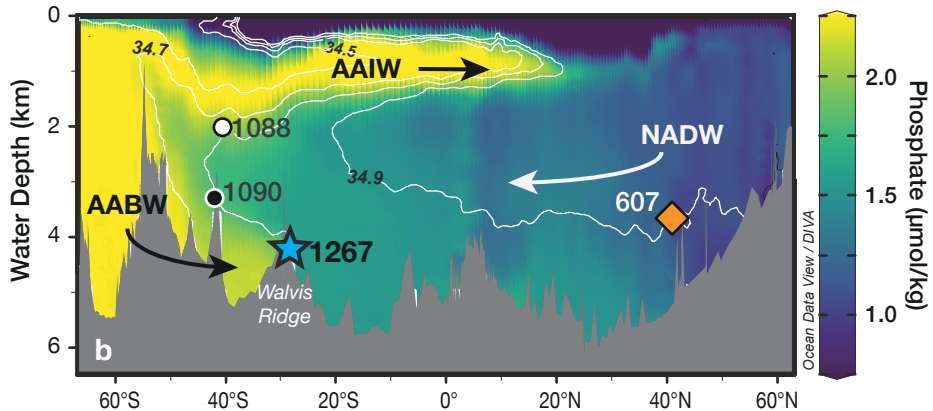
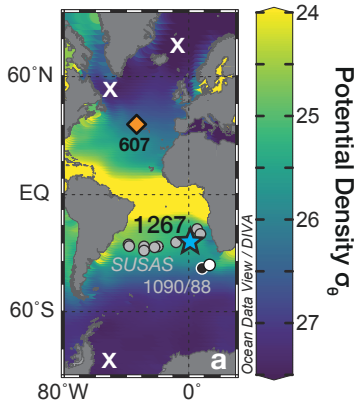
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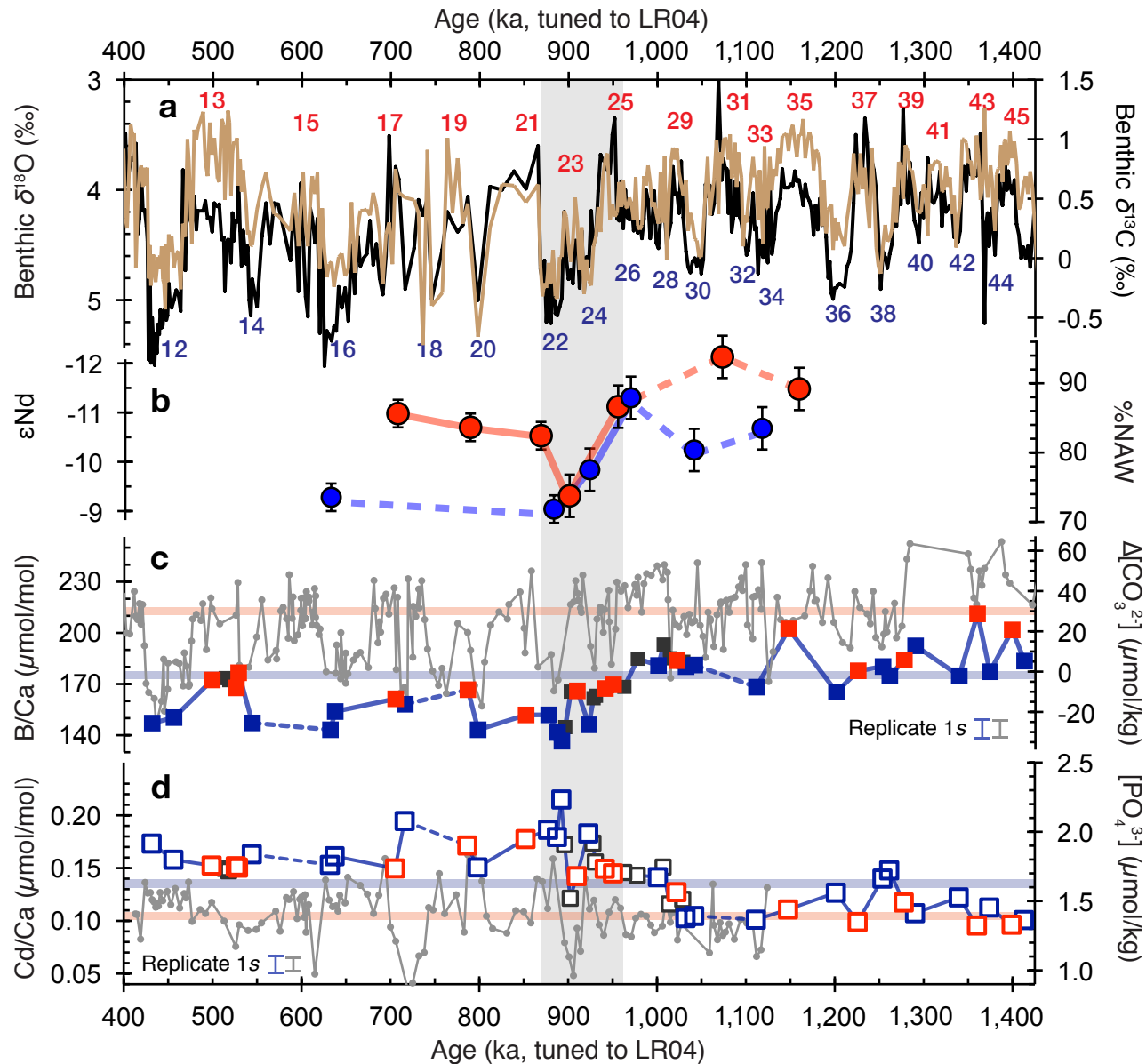
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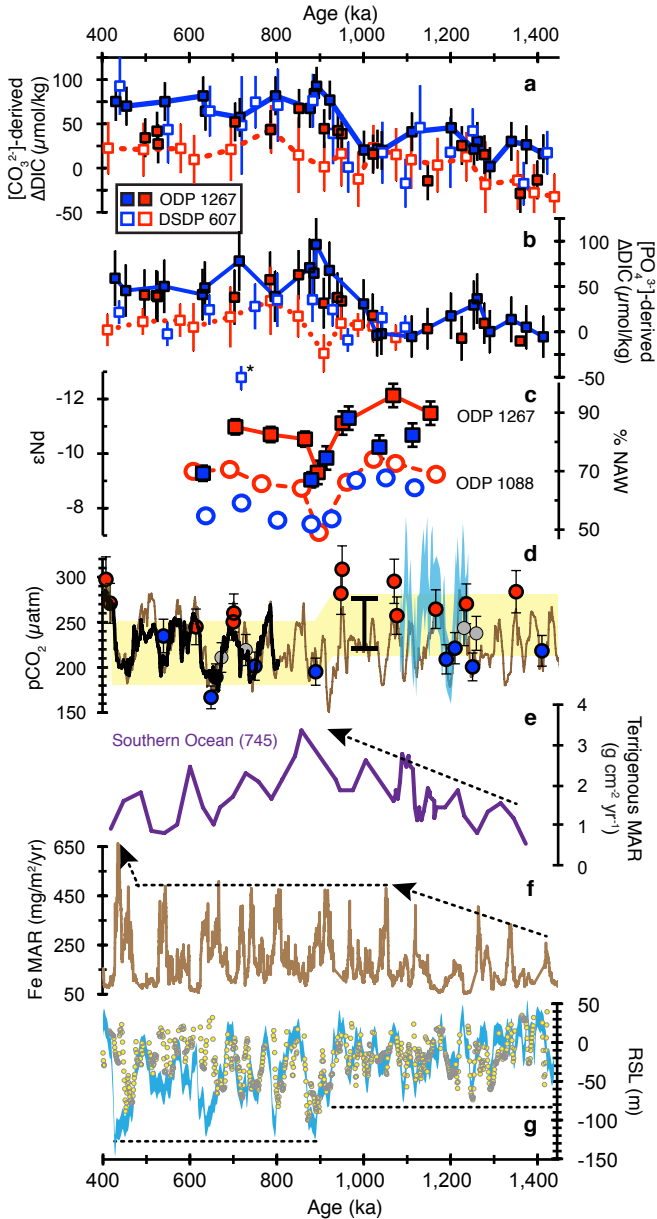
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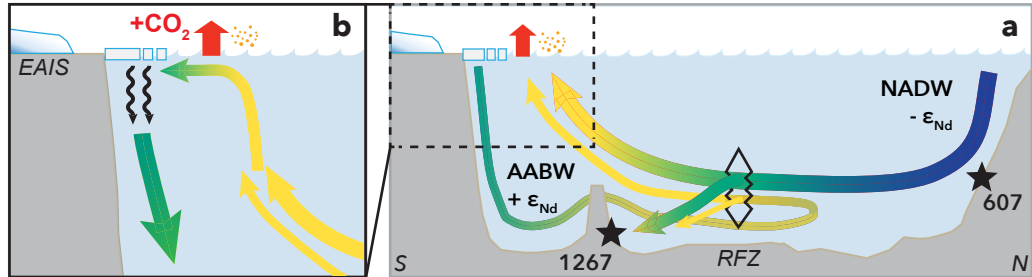
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Pre-MPT Glacials (1,400 to 1,000 ka)



Post-MPT Glacials (after 950 ka)

